


3.00 credits

30.0 h

Q1

|                             |  |
|-----------------------------|--|
| Teacher(s)                  | Dupont Christine ;Garcia Yann ;  |
| Language :                  | French   |
| Place of the course         | Louvain-la-Neuve   |
| Prerequisites               | It is recommended to have acquired the knowledge and skills developed in the teaching units:<br><a href="#">LCHM1111</a> Chimie générale<br><a href="#">LCHM1211</a> Chimie générale 2<br><a href="#">LPHY1101</a> Physique 1<br><a href="#">LPHY1102</a> Physique 2   |
| Main themes                 | At first, the course brings the student to a good knowledge in solution thermodynamics and to the quantitative prediction of their behaviour. Activity and standard state concepts must be used in a reasonable way at this stage. The different classes of reactions are then developed in order to rigorously exploit basic operations in quantitative chemical analysis. The study of gravimetry and titration allows illustrating fundamental bases of operating modes. Theoretical aspects of chromatographic separation methods as well as an introduction to spectrochemical analysis are given. Finally, theoretical bases and applications of potentiometry to an analytical problem are described. The student is here sensitised to important concepts such as electrode potentials, reference electrode, indicator electrode, and to the correspondence of an electrochemical circuit to the needs of analysis as well as analytical performances. The care specific to potentiometric methods is also outlined. |
| Learning outcomes           |  |
| Evaluation methods          | Tests during the semester (20%) - Written exam (80%)   |
| Teaching methods            | Lectures / exercise sessions (after preparation of exercises made available online).<br>Some classes could be given remotely depending on the pandemic situation.  |
| Content                     | <p>Introduction</p> <ul style="list-style-type: none"> <li>- Chemical analysis and information (analysis performances, experimental error and its treatment)</li> <li>- Aqueous solutions of electrolytes (ionic strength, activity coefficient, chemical potential)</li> </ul> <p>Introduction to spectroscopy (phenomena, devices, quantitative exploitation)</p> <ul style="list-style-type: none"> <li>- precipitation and gravimetry</li> <li>- volumetry</li> <li>- redox reactions</li> <li>- potentiometry (indicator and reference electrodes, membrane potential, complex sensors)</li> </ul> <p>Introduction to chromatography (theory)</p>   |
| Inline resources            | LCHM1321 Moodle website and LCHM1321 Teams   |
| Bibliography                | Skoog and West   |
| Faculty or entity in charge | CHIM   |

| Programmes containing this learning unit (UE)               |         |         |              |   |
|---|---------|---------|--------------|---|
| Program title   | Acronym | Credits | Prerequisite | Learning outcomes   |
| Master [120] in Biochemistry and Molecular and Cell Biology | BBMC2M  | 3       |              |  |