


6.00 credits

30.0 h + 45.0 h

Q1

Teacher(s)	Hermans Sophie ;
Language :	English > French-friendly
Place of the course	Louvain-la-Neuve
Main themes	<p>The course will concern the fundamental concepts in coordination chemistry considering electronic aspects (spectra and magnetic properties), structural aspects (isolobal analogy) and reactivity (reaction mechanisms).</p> <p>Practical exercises will cover :</p> <ul style="list-style-type: none"> <li>- the synthesis and purification of transition metal coordination compounds</li> <li>- the mastery of principal characterisation technique (principally spectroscopic) of inorganic compounds.</li> </ul>
Learning outcomes	<p><b>At the end of this learning unit, the student is able to :</b></p> <p>This course aims at covering the principal basic concepts in coordination chemistry. Practical exercises will concern the synthesis and study of physico-chemical properties of transition metals coordination compounds.</p> <p>1 <b>Prerequisite :</b></p> <ul style="list-style-type: none"> <li>Molecular symmetry and crystal structures (CHM 1251A).</li> <li>Basics of molecular spectroscopy (CHM 1251B).</li> <li>Booklet including the copy of the transparencies used by the teacher</li> </ul>
Evaluation methods	<p>The examination is a written + oral exam during the session.</p> <p>The final note also comprises a contribution from the evaluation of practical labs (synthesized products + written reports)</p>
Teaching methods	Theoretical course in auditorium and practical laboratories in teaching labs.
Content	<p>The course will cover the following aspects :</p> <ol style="list-style-type: none"> <li>1. General properties of coordination compounds: electronic spectroscopy and magnetic properties, description of bonding in the framework of the molecular orbital theory.</li> <li>2. Reaction mechanisms in coordination chemistry. Ligand substitution reactions (octahedral complexes, square-planar complexes: trans effect). Electron transfer reactions.</li> <li>3. Organometallic chemistry complements: isolobal analogies.</li> <li>4. Molecular polyhedra in inorganic chemistry: the metal-metal bond, boranes structure, metallic clusters.</li> <li>5. Bioinorganic chemistry.</li> </ol> <p>The practical laboratories cover manipulations among the following themes :</p> <ol style="list-style-type: none"> <li>1. Synthesis and spectroscopic characterisation of Vanadium complexes.</li> <li>2. Synthesis and spectroscopic characterisation of Cr(III) complexes.</li> <li>3. Synthesis and spectroscopic characterisation of Ni(II) complexes.</li> <li>4. Synthesis of luminescent compounds.</li> <li>5. Separation of optical isomers of Co(III) complexes.</li> <li>6. Kinetics of cis-trans isomerization.</li> <li>7. The Job method.</li> <li>8. Ambidentates ligands and linkage isomerism.</li> </ol>
Inline resources	The visual support used by the teacher is available on Moodle.
Bibliography	<p>Supports :</p> <ul style="list-style-type: none"> <li>- "Inorganic Chemistry : principles of structure and reactivity", J. Huheey, E. Keiter, R. Keiter, 4th ed., Harper and Collins, 1993.</li> <li>- Autres références bibliographiques conseillées au début de l'enseignement.</li> <li>- Copie des transparents utilisés par l'enseignant, disponibles sur Moodle.</li> <li>- Pour les exercices pratiques : manuel de laboratoire</li> </ul>

Other infos	<p><b>Background :</b>                  General chemistry notions                  Molecular symmetry and crystal structures                  Fundamentals of theoretical chemistry and molecular spectroscopy.                  Inorganic chemistry I (CHM 1331).</p>
Faculty or entity in charge	CHIM

<b>Programmes containing this learning unit (UE)</b>				
Program title	Acronym	Credits	Prerequisite	Learning outcomes
Master [120] in Chemistry	<a href="#">CHIM2M</a>	6		
Master [60] in Chemistry	<a href="#">CHIM2M1</a>	6		