UCLouvain

lphys2246

2025

Experimental methods in atomic and molecular physics

The version you're consulting is not final. This course description may change. The final version will be published on 1st June.

5.00 credits	30.0 h	Q2
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themes of this teaching unit are charge particle optics, atomic and electronic collisions, and molecular spectroscopy. ate on the means to produce, store and guide charged particles using electric and magnetic illustrate the relevance of this know-how to the study of cross sections of collisions or photonocesses. An emphasis is then put on ultra-sensitive and precise techniques of spectroscopy detection of photons or of charged particles. Different cooling techniques, i.e. supersonic and buffer gas cooling, are also presented to simplify and enhance quantized signatures in or collision experiments. To this learning unit, the student is able to: Contribution of the teaching unit to the learning outcomes of the programme (PHYS2M PHYS2M1) 1.1, AA 1.2, AA1.3, AA1.4, AA 1.5, AA1.6, AA2.1, AA2.2, AA 3.1, AA 4.2, AA5.1, AA5.2, AA AA 6.1, AA 7.2, AA 7.3, AA7.5, AA8.1, AA 8.2 Specific learning outcomes of the teaching unit
I molecular spectroscopy. ate on the means to produce, store and guide charged particles using electric and magnetic illustrate the relevance of this know-how to the study of cross sections of collisions or photon-ocesses. An emphasis is then put on ultra-sensitive and precise techniques of spectroscopy detection of photons or of charged particles. Different cooling techniques, i.e. supersonic and buffer gas cooling, are also presented to simplify and enhance quantized signatures in or collision experiments. Of this learning unit, the student is able to: Contribution of the teaching unit to the learning outcomes of the programme (PHYS2M PHYS2M1) 1.1, AA 1.2, AA1.3, AA1.4, AA 1.5, AA1.6, AA2.1, AA2.2, AA 3.1, AA 4.2, AA5.1, AA5.2, AA AA 6.1, AA 7.2, AA 7.3, AA7.5, AA8.1, AA 8.2 Specific learning outcomes of the teaching unit
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ne end of this teaching unit, the student will be able to: determine the most efficient experimental methodology to study a problem in atomic or ecular physics; know what are the limitations and advantages of various experimental techniques in atomic molecular physics; identify the methods in use in scientific publications and evaluate their pertinence put into equations the trajectory of charged particle beam and simulate it with appropriate ware tools; identify and characterize the elements of a particle accelerator.
ation will be based on an individual project and its oral presentation.
aboratories, practical project, commented laboratory tours.
Ing unit will adopt the following structure: If particle optics generation of charged particles: electron, positron, ion basic principles of charged particle optics: general equations of motion, paraxial approximation and applications to electric and magnetic fields concept of emittance: Liouville theorem and derivation of the beam envelope in phase space practical training with real beams and simulation tools ental approach of atomic and electronic collisions velocity distributions: gas cell, effusive and supersonic beam velocity selection: rotating slit, Doppler, fast beam

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	• frequency modulation • -principle of a lock-in amplifier • cavity enhanced and cavity ringdown spectroscopy • NICE-OHMS spectroscopy
	Action spectroscopy
	• photofragmentation spectroscopy
	• photoelectron spectroscopy
	spectroscopy in an ion-trap
	The teaching unit will incorporate the latest experimental developments in atomic and molecular physics.
	Visits to a large European experimental facility will be organised.
Bibliography	H. Wollnik, Optics of Charged Particles (Academic Press, Orlando, 1987).
	High-resolution molecular spectroscopy, handbook, Wiley online library 2011.
Faculty or entity in charge	PHYS

Programmes containing this learning unit (UE)					
Program title	Acronym	Credits	Prerequisite	Learning outcomes	
Master [60] in Physics	PHYS2M1	5		•	
Master [120] in Physical Engineering	FYAP2M	5		•	
Master [120] in Physics	PHYS2M	5		Q	